

(1S,3R,8R,9S,11R)-10,10-Dibromo-2,2-dichloro-3,7,7,11-tetramethyltetracyclo-[6.5.0.0^{1,3}.0^{9,11}]tridecane

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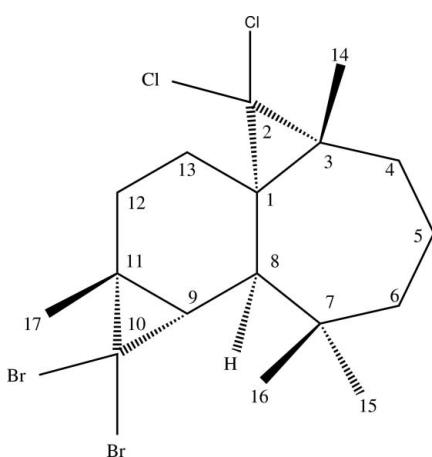
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Key indicators: single-crystal X-ray study; $T = 296\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$; R factor = 0.041; wR factor = 0.103; data-to-parameter ratio = 24.2.

The title compound, $C_{17}H_{24}Br_2Cl_2$, was synthesized from β -himachalene (3,5,5,9-tetramethyl-2,4a,5,6,7,8-hexahydro-1*H*-benzocycloheptene), which was isolated from the essential oil of the Atlas cedar (*Cedrus Atlantica*). The asymmetric unit contains two independent molecules. Each molecule is built up from fused six-, seven- and two three-membered rings. In both molecules, the six-membered ring has a half-chair conformation, whereas the seven-membered ring displays a boat conformation. No specific intermolecular interactions are noted in the crystal packing.

Related literature

For similar compounds, see: Ourhriss *et al.* (2013); Oukhrib *et al.* (2013a,b). For the biological properties of β -himachalene, see: El Haib *et al.* (2011). For puckering parameters, see: Cremer & Pople (1975).



Experimental

Crystal data

$C_{17}H_{24}Br_2Cl_2$	$V = 3687 (6)\text{ \AA}^3$
$M_r = 459.08$	$Z = 8$
Monoclinic, $C2$	Mo $K\alpha$ radiation
$a = 18.377 (16)\text{ \AA}$	$\mu = 4.68\text{ mm}^{-1}$
$b = 6.519 (6)\text{ \AA}$	$T = 296\text{ K}$
$c = 30.82 (3)\text{ \AA}$	$0.43 \times 0.31 \times 0.27\text{ mm}$
$\beta = 93.233 (16)^\circ$	

Data collection

Bruker X8 APEX diffractometer	40489 measured reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2009)	9155 independent reflections
$(SADABS$; Bruker, 2009)	6959 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.739$, $T_{\max} = 0.867$	$R_{\text{int}} = 0.037$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$	$\Delta\rho_{\max} = 0.67\text{ e \AA}^{-3}$
$wR(F^2) = 0.103$	$\Delta\rho_{\min} = -0.44\text{ e \AA}^{-3}$
$S = 1.04$	Absolute structure: Flack &
9155 reflections	Bernardinelli (2000), 3988 Friedel pairs
379 parameters	Flack parameter: 0.029 (8)
1 restraint	
H-atom parameters constrained	

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *PLATON* (Spek, 2009) and *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: VM2192).

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supplementary materials

Acta Cryst. (2013). E69, o739 [doi:10.1107/S1600536813010040]

(1*S*,3*R*,8*R*,9*S*,11*R*)-10,10-Dibromo-2,2-dichloro-3,7,7,11-tetramethyltetracyclo-[6.5.0.0^{1,3}.0^{9,11}]tridecane

Abdelouahd Oukhrib, Ahmed Benharref, Mohamed Saadi, Moha Berraho and Lahcen El Ammani

Comment

This work is a part of our ongoing program concerning the valorization of the most abundant essential oils in Morocco, such as *Cedrus atlantica*. This oil is made up mainly (50%) of β -himachalene (Ourhriss *et al.*, 2013; Oukhrib *et al.*, 2013*a,b*). The reactivity of this sesquiterpene and its derivatives has been studied extensively by our team in order to prepare new products having biological properties (El Haib *et al.*, 2011). We present here the crystal structure of the title compound, (1*S*,3*R*,8*R*,9*S*,11*R*)-2,2-dichloro-10,10-dibromo-3,7,7,11-tetramethyltetracyclo [6.5.0.0^{1,3}.0^{9,11}]tridecane noted (I).

As shown in Fig. 1, each molecule of the title compound is build up by fused six- and seven-membered rings, which are fused to two three-membered rings. In the first molecule, the total puckering amplitude QT = 0.462 (4) Å and spherical polar angle θ_2 = 142.4 (5) $^\circ$ and φ_2 = 136.2 (8) $^\circ$, obtained for the six-membered ring are compatible with a half chair conformation, whereas those calculated for the seven-membered ring (QT = 1.146 (5) Å, θ_2 = 87.1 (3) $^\circ$, φ_2 = -48.8 (3) $^\circ$ and φ_3 = -122 (4) $^\circ$ (Cremer & Pople, 1975)) involve a boat conformation. Nearly the same values are observed in the second molecule, and consequently the same conformation is present, as shown in the fitting drawing (Fig. 2, r.m.s. deviation 0.057 Å). Owing to the presence of Br and Cl atoms, the absolute configuration could be fully confirmed from anomalous dispersion effects, by refining the Flack parameter as C1(S),C3(R),C8(R),C9(S), and C11(R). The crystal packing exhibits no short intermolecular contacts.

Experimental

A solution containing 3 g (8 mmol) of (1*S*,3*R*,8*S*)-2,2-dichloro- 3,7,7,10- tetramethyltricyclo [6.4.0.0^{1,3}]dodec-9-ene and 1 ml (10 mmol) of CHBr₃ in 40 ml of dichloromethane was added dropwise at 273 K over 30 min to 1 g of pulverized sodium hydroxide and 40 mg of *N*-benzyltriethylammonium chloride placed in a 100 ml three-necked flask. After stirring at room temperature for 2 h, the mixture was filtered on celite and concentrated in vacuum. The residue obtained was chromatographed on silicagel column impregnated with silver nitrate (10%) with a mixture of hexane - ethyl acetate (96–4) used as eluent. The two diastereoisomers (1*S*,3*R*,8*R*,9*S*,11*R*)-10,10-dichloro-2,2-dibromo-3,7,7,11- tetramethyltetracyclo [6.5.0.0^{1,3}.0^{9,11}]tridecane (*X*) and its isomer (1*S*,3*R*,8*R*, 9*R*, 11*S*)- 10,10- dichloro-2,2-dibromo-3,7,7,11-tetramethyltetracyclo[6.5.0.0^{1,3}.0^{9,11}] tridecane (*Y*), were obtained by this procedure in a 85/15 ratio and a combined yield of 70% (2.5 g; 5.6 mmol). The title compound (isomer *X*) was recrystallized from hexane.

Refinement

All H atoms were fixed geometrically and treated as riding with C—H = 0.96 Å (methyl), 0.97 Å (methylene), 0.98 Å (methine) with $U_{\text{iso}}(\text{H})$ = 1.2 U_{eq} (methylene, methine) and $U_{\text{iso}}(\text{H})$ = 1.5 U_{eq} (methyl). The space group is

noncentrosymmetric and the polar axis restraint is generated automatically by *SHELXL* program. The 3988 Friedel pairs were not merged.

Computing details

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT* (Bruker, 2009); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *PLATON* (Spek, 2009) and *publCIF* (Westrip, 2010).

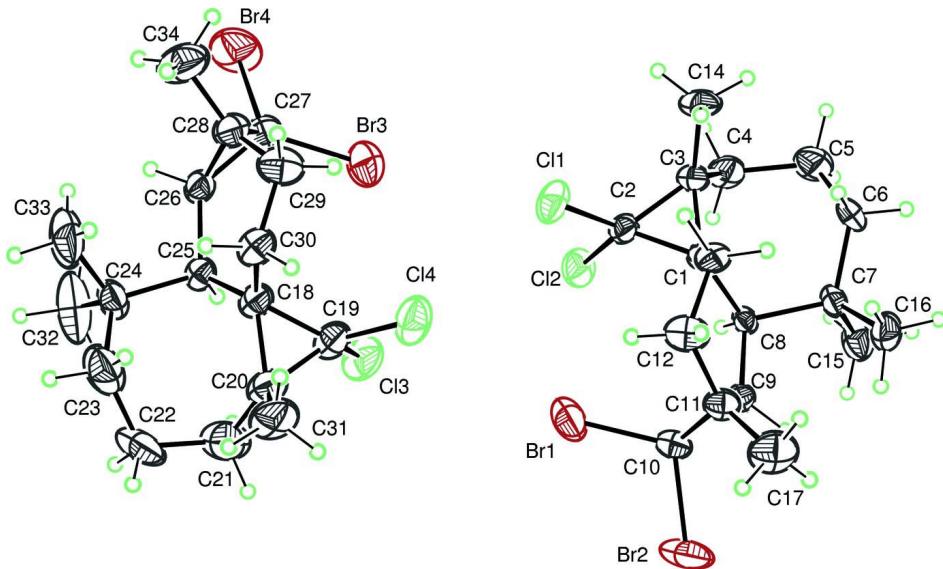
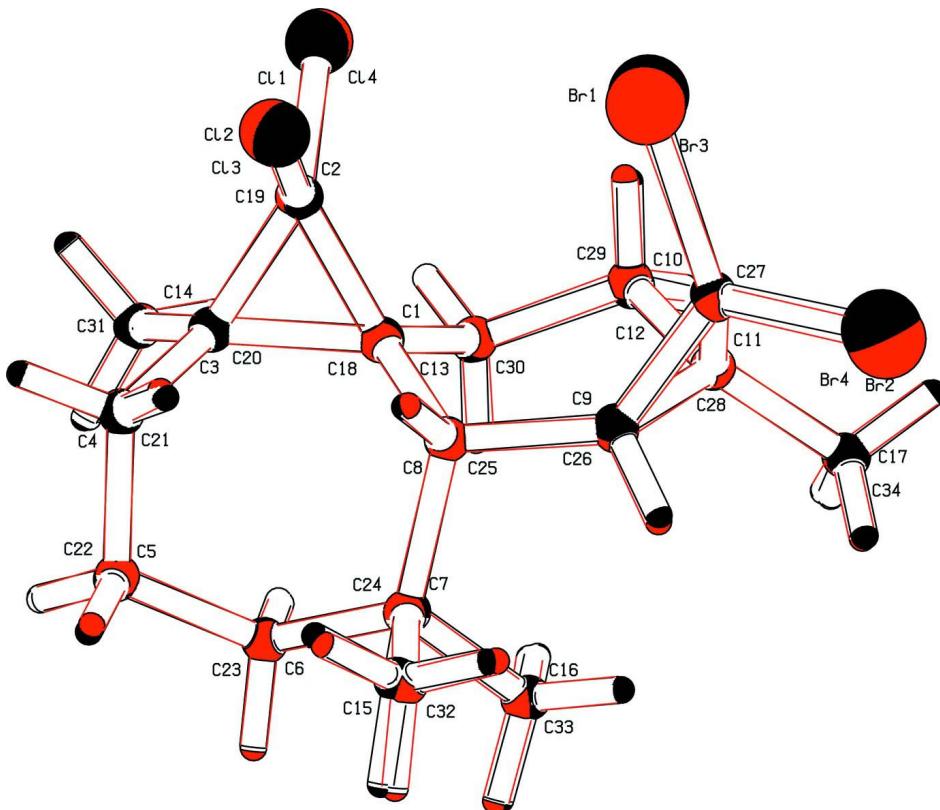


Figure 1

Molecular structure of the title compound with the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are represented as small spheres of arbitrary radii.

**Figure 2**

View showing the fitting of the two molecules building the asymmetric unit.

(1*S*,3*R*,8*R*,9*S*,11*R*)-10,10-Bibromo-2,2-dichloro-3,7,7,11-tetramethyltetracyclo[6.5.0.0^{1,3}.0^{9,11}]tridecane

Crystal data



M_r = 459.08

Monoclinic, C2

Hall symbol: c 2y

a = 18.377 (16) Å

b = 6.519 (6) Å

c = 30.82 (3) Å

β = 93.233 (16)°

V = 3687 (6) Å³

Z = 8

F(000) = 1840

D_x = 1.654 Mg m⁻³

Mo Kα radiation, λ = 0.71073 Å

Cell parameters from 9155 reflections

θ = 2.4–28.7°

μ = 4.68 mm⁻¹

T = 296 K

Block, colourless

0.43 × 0.31 × 0.27 mm

Data collection

Bruker X8 APEX

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan

(SADABS; Bruker, 2009)

T_{min} = 0.739, T_{max} = 0.867

40489 measured reflections

9155 independent reflections

6959 reflections with *I* > 2σ(*I*)

R_{int} = 0.037

θ_{max} = 28.7°, θ_{min} = 2.4°

h = -24→24

k = -8→8

l = -41→41

*Refinement*Refinement on F^2

Least-squares matrix: full

$$R[F^2 > 2\sigma(F^2)] = 0.041$$

$$wR(F^2) = 0.103$$

$$S = 1.04$$

9155 reflections

379 parameters

1 restraint

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0444P)^2 + 4.0898P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} = 0.001$$

$$\Delta\rho_{\max} = 0.67 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.44 \text{ e } \text{\AA}^{-3}$$

Absolute structure: Flack & Bernardinelli
(2000), 3988 Friedel pairs

Flack parameter: 0.029 (8)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against all reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on all data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Br1	0.69402 (2)	0.22377 (9)	0.666261 (16)	0.05626 (15)
Br2	0.63800 (2)	0.29919 (8)	0.56956 (2)	0.06228 (17)
Br3	0.81097 (3)	0.09644 (10)	0.840122 (16)	0.06037 (16)
Br4	0.85782 (3)	0.15293 (10)	0.93871 (2)	0.06974 (19)
Cl4	0.66339 (7)	-0.1805 (2)	0.76362 (3)	0.0535 (3)
Cl3	0.65444 (7)	0.24769 (19)	0.78504 (4)	0.0510 (3)
Cl2	0.85390 (6)	0.41099 (15)	0.71165 (3)	0.0384 (2)
C11	0.85215 (6)	-0.01541 (18)	0.73567 (3)	0.0441 (3)
C8	0.86557 (16)	0.2398 (5)	0.61173 (10)	0.0204 (6)
H8	0.8641	0.3786	0.6239	0.024*
C7	0.92400 (19)	0.2407 (6)	0.57466 (11)	0.0259 (7)
C1	0.88476 (18)	0.0909 (5)	0.64962 (10)	0.0213 (7)
C2	0.8812 (2)	0.1600 (6)	0.69673 (11)	0.0281 (8)
C9	0.79025 (18)	0.1957 (5)	0.58910 (11)	0.0244 (7)
H9	0.7874	0.2475	0.5592	0.029*
C10	0.71591 (19)	0.1842 (6)	0.60708 (12)	0.0315 (8)
C13	0.8639 (2)	-0.1306 (5)	0.63885 (12)	0.0279 (8)
H13A	0.8783	-0.2171	0.6635	0.034*
H13B	0.8906	-0.1759	0.6143	0.034*
C12	0.7817 (2)	-0.1575 (6)	0.62784 (15)	0.0391 (10)
H12A	0.7726	-0.2978	0.6186	0.047*
H12B	0.7557	-0.1345	0.6539	0.047*
C3	0.95537 (19)	0.1338 (6)	0.67808 (12)	0.0289 (8)
C11	0.7515 (2)	-0.0120 (6)	0.59218 (12)	0.0315 (8)

C4	0.9979 (2)	0.3277 (7)	0.66790 (12)	0.0362 (9)
H4A	1.0303	0.3626	0.6927	0.043*
H4B	0.9641	0.4406	0.6628	0.043*
C16	0.9041 (2)	0.0880 (7)	0.53687 (12)	0.0400 (10)
H16A	0.9027	-0.0491	0.5482	0.060*
H16B	0.9402	0.0961	0.5156	0.060*
H16C	0.8572	0.1230	0.5236	0.060*
C15	0.9259 (2)	0.4560 (7)	0.55567 (14)	0.0439 (10)
H15A	0.8777	0.4953	0.5453	0.066*
H15B	0.9578	0.4580	0.5320	0.066*
H15C	0.9436	0.5504	0.5777	0.066*
C5	1.0427 (2)	0.2985 (8)	0.62777 (14)	0.0465 (11)
H5A	1.0877	0.2277	0.6364	0.056*
H5B	1.0554	0.4321	0.6166	0.056*
C6	1.00158 (19)	0.1757 (7)	0.59140 (13)	0.0376 (10)
H6A	0.9986	0.0349	0.6013	0.045*
H6B	1.0318	0.1751	0.5666	0.045*
C24	0.5723 (2)	0.0880 (7)	0.92056 (13)	0.0387 (10)
C25	0.63553 (19)	0.0851 (6)	0.88610 (11)	0.0276 (8)
H25	0.6389	0.2233	0.8738	0.033*
C28	0.7494 (2)	-0.1594 (7)	0.90993 (13)	0.0382 (9)
C14	1.0047 (2)	-0.0449 (8)	0.69381 (14)	0.0454 (11)
H14A	1.0493	-0.0406	0.6791	0.068*
H14B	0.9803	-0.1726	0.6876	0.068*
H14C	1.0152	-0.0332	0.7246	0.068*
C18	0.6206 (2)	-0.0658 (6)	0.84803 (12)	0.0319 (9)
C20	0.5526 (2)	-0.0342 (8)	0.81676 (13)	0.0433 (10)
C26	0.7074 (2)	0.0406 (6)	0.91218 (12)	0.0330 (9)
H26	0.7057	0.0888	0.9422	0.040*
C34	0.7739 (3)	-0.2710 (10)	0.95177 (16)	0.0646 (15)
H34A	0.7455	-0.3934	0.9545	0.097*
H34B	0.7672	-0.1830	0.9762	0.097*
H34C	0.8245	-0.3067	0.9510	0.097*
C30	0.6433 (2)	-0.2858 (7)	0.85972 (13)	0.0369 (9)
H30A	0.6156	-0.3328	0.8837	0.044*
H30B	0.6315	-0.3744	0.8350	0.044*
C17	0.7213 (3)	-0.1179 (8)	0.55062 (16)	0.0562 (13)
H17A	0.7494	-0.2389	0.5456	0.084*
H17B	0.7242	-0.0259	0.5265	0.084*
H17C	0.6714	-0.1554	0.5538	0.084*
C21	0.5072 (3)	0.1560 (10)	0.82436 (19)	0.0658 (16)
H21A	0.5396	0.2727	0.8287	0.079*
H21B	0.4759	0.1826	0.7986	0.079*
C33	0.5880 (3)	-0.0637 (10)	0.95866 (17)	0.0746 (18)
H33A	0.5898	-0.2011	0.9475	0.112*
H33B	0.5500	-0.0537	0.9787	0.112*
H33C	0.6339	-0.0303	0.9733	0.112*
C19	0.6278 (2)	-0.0030 (7)	0.80028 (13)	0.0379 (9)
C27	0.7848 (2)	0.0368 (7)	0.89782 (13)	0.0385 (10)

C29	0.7244 (2)	-0.3040 (7)	0.87242 (15)	0.0448 (10)
H29A	0.7519	-0.2733	0.8473	0.054*
H29B	0.7351	-0.4444	0.8810	0.054*
C22	0.4606 (2)	0.1369 (10)	0.86313 (18)	0.0636 (15)
H22A	0.4155	0.0676	0.8540	0.076*
H22B	0.4481	0.2733	0.8729	0.076*
C32	0.5706 (4)	0.2969 (9)	0.94119 (19)	0.079 (2)
H32A	0.6188	0.3333	0.9523	0.118*
H32B	0.5383	0.2949	0.9646	0.118*
H32C	0.5538	0.3960	0.9199	0.118*
C31	0.5073 (3)	-0.2163 (9)	0.80026 (16)	0.0604 (14)
H31A	0.4612	-0.2156	0.8135	0.091*
H31B	0.5328	-0.3413	0.8076	0.091*
H31C	0.4994	-0.2071	0.7693	0.091*
C23	0.4979 (3)	0.0178 (10)	0.90194 (17)	0.0654 (16)
H23A	0.4649	0.0196	0.9254	0.078*
H23B	0.5026	-0.1241	0.8930	0.078*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Br1	0.0364 (2)	0.0805 (4)	0.0533 (3)	0.0029 (2)	0.0148 (2)	-0.0100 (3)
Br2	0.0302 (2)	0.0682 (4)	0.0871 (4)	0.0114 (2)	-0.0092 (2)	0.0232 (3)
Br3	0.0448 (3)	0.0867 (4)	0.0511 (3)	-0.0068 (3)	0.0154 (2)	0.0094 (3)
Br4	0.0436 (3)	0.0861 (5)	0.0779 (4)	-0.0139 (3)	-0.0117 (2)	-0.0230 (3)
Cl4	0.0625 (7)	0.0642 (8)	0.0342 (6)	-0.0114 (6)	0.0055 (5)	-0.0110 (5)
Cl3	0.0611 (7)	0.0489 (7)	0.0432 (6)	-0.0064 (5)	0.0051 (5)	0.0183 (5)
Cl2	0.0467 (6)	0.0344 (5)	0.0346 (5)	0.0058 (4)	0.0054 (4)	-0.0094 (4)
C11	0.0583 (6)	0.0463 (6)	0.0285 (5)	-0.0011 (5)	0.0087 (4)	0.0104 (4)
C8	0.0196 (15)	0.0197 (16)	0.0222 (16)	0.0038 (12)	0.0046 (12)	0.0001 (14)
C7	0.0248 (17)	0.0261 (19)	0.0275 (17)	0.0037 (14)	0.0085 (13)	0.0001 (15)
C1	0.0224 (16)	0.0195 (16)	0.0221 (16)	0.0050 (13)	0.0008 (13)	0.0007 (14)
C2	0.0334 (19)	0.0292 (19)	0.0216 (17)	0.0083 (15)	-0.0001 (14)	0.0016 (15)
C9	0.0252 (16)	0.0254 (18)	0.0228 (16)	0.0053 (14)	0.0029 (13)	0.0052 (14)
C10	0.0220 (17)	0.034 (2)	0.038 (2)	0.0016 (15)	-0.0021 (15)	0.0017 (17)
C13	0.036 (2)	0.0145 (16)	0.033 (2)	0.0073 (14)	-0.0022 (15)	0.0042 (14)
C12	0.038 (2)	0.024 (2)	0.055 (3)	-0.0049 (17)	-0.0053 (19)	0.0035 (19)
C3	0.0243 (17)	0.034 (2)	0.0276 (18)	0.0054 (15)	-0.0025 (14)	-0.0004 (16)
C11	0.0299 (19)	0.0282 (19)	0.036 (2)	-0.0007 (16)	-0.0033 (15)	-0.0016 (17)
C4	0.0292 (19)	0.042 (2)	0.037 (2)	-0.0100 (18)	-0.0069 (16)	-0.0043 (18)
C16	0.046 (2)	0.045 (2)	0.029 (2)	0.008 (2)	0.0080 (17)	-0.0047 (19)
C15	0.050 (3)	0.040 (2)	0.044 (2)	0.000 (2)	0.019 (2)	0.009 (2)
C5	0.0259 (19)	0.064 (3)	0.049 (2)	-0.008 (2)	-0.0007 (17)	0.006 (2)
C6	0.0231 (18)	0.054 (3)	0.036 (2)	0.0111 (18)	0.0109 (15)	0.0012 (19)
C24	0.030 (2)	0.048 (3)	0.040 (2)	-0.0042 (18)	0.0111 (16)	-0.005 (2)
C25	0.0291 (18)	0.0270 (19)	0.0267 (18)	-0.0070 (15)	0.0024 (14)	0.0031 (16)
C28	0.034 (2)	0.045 (3)	0.035 (2)	-0.0004 (19)	-0.0016 (17)	0.0050 (19)
C14	0.039 (2)	0.051 (3)	0.044 (2)	0.020 (2)	-0.0100 (19)	0.005 (2)
C18	0.0308 (19)	0.037 (2)	0.0280 (19)	-0.0081 (16)	-0.0006 (15)	0.0040 (17)
C20	0.038 (2)	0.055 (3)	0.036 (2)	-0.008 (2)	-0.0061 (18)	0.007 (2)

C26	0.0283 (19)	0.046 (2)	0.0245 (18)	-0.0060 (17)	0.0006 (15)	-0.0008 (17)
C34	0.066 (3)	0.069 (4)	0.057 (3)	0.001 (3)	-0.012 (3)	0.023 (3)
C30	0.044 (2)	0.032 (2)	0.035 (2)	-0.0077 (18)	-0.0008 (17)	0.0036 (18)
C17	0.051 (3)	0.055 (3)	0.061 (3)	-0.012 (2)	-0.012 (2)	-0.011 (3)
C21	0.046 (3)	0.076 (4)	0.075 (4)	0.005 (3)	0.000 (3)	0.026 (3)
C33	0.088 (4)	0.075 (4)	0.065 (3)	0.007 (3)	0.048 (3)	0.020 (3)
C19	0.042 (2)	0.042 (2)	0.030 (2)	-0.0063 (19)	-0.0020 (16)	0.0045 (18)
C27	0.033 (2)	0.050 (3)	0.032 (2)	-0.0042 (18)	-0.0054 (16)	-0.0078 (19)
C29	0.047 (2)	0.027 (2)	0.060 (3)	0.0029 (19)	-0.005 (2)	-0.002 (2)
C22	0.028 (2)	0.072 (4)	0.091 (4)	0.011 (2)	0.007 (2)	0.008 (3)
C32	0.108 (5)	0.056 (3)	0.079 (4)	-0.024 (3)	0.060 (4)	-0.020 (3)
C31	0.045 (3)	0.080 (4)	0.055 (3)	-0.023 (3)	-0.015 (2)	-0.001 (3)
C23	0.044 (3)	0.083 (4)	0.071 (4)	-0.007 (3)	0.021 (3)	-0.007 (3)

Geometric parameters (\AA , $^\circ$)

Br1—C10	1.907 (4)	C24—C33	1.550 (7)
Br2—C10	1.940 (4)	C24—C25	1.618 (5)
Br3—C27	1.908 (4)	C25—C26	1.535 (5)
Br4—C27	1.942 (4)	C25—C18	1.544 (5)
Cl4—C19	1.768 (5)	C25—H25	0.9800
Cl3—C19	1.777 (5)	C28—C27	1.491 (6)
Cl2—C2	1.779 (4)	C28—C26	1.518 (6)
Cl1—C2	1.762 (4)	C28—C34	1.526 (6)
C8—C9	1.541 (5)	C28—C29	1.542 (6)
C8—C1	1.544 (5)	C14—H14A	0.9600
C8—C7	1.612 (4)	C14—H14B	0.9600
C8—H8	0.9800	C14—H14C	0.9600
C7—C15	1.522 (6)	C18—C30	1.531 (6)
C7—C6	1.548 (5)	C18—C19	1.541 (5)
C7—C16	1.560 (5)	C18—C20	1.549 (5)
C1—C2	1.525 (5)	C20—C19	1.514 (6)
C1—C13	1.526 (5)	C20—C31	1.520 (7)
C1—C3	1.550 (5)	C20—C21	1.520 (8)
C2—C3	1.519 (5)	C26—C27	1.513 (6)
C9—C10	1.505 (5)	C26—H26	0.9800
C9—C11	1.535 (5)	C34—H34A	0.9600
C9—H9	0.9800	C34—H34B	0.9600
C10—C11	1.519 (6)	C34—H34C	0.9600
C13—C12	1.539 (5)	C30—C29	1.523 (6)
C13—H13A	0.9700	C30—H30A	0.9700
C13—H13B	0.9700	C30—H30B	0.9700
C12—C11	1.532 (6)	C17—H17A	0.9600
C12—H12A	0.9700	C17—H17B	0.9600
C12—H12B	0.9700	C17—H17C	0.9600
C3—C4	1.528 (6)	C21—C22	1.515 (7)
C3—C14	1.537 (5)	C21—H21A	0.9700
C11—C17	1.532 (6)	C21—H21B	0.9700
C4—C5	1.536 (6)	C33—H33A	0.9600
C4—H4A	0.9700	C33—H33B	0.9600

C4—H4B	0.9700	C33—H33C	0.9600
C16—H16A	0.9600	C29—H29A	0.9700
C16—H16B	0.9600	C29—H29B	0.9700
C16—H16C	0.9600	C22—C23	1.553 (8)
C15—H15A	0.9600	C22—H22A	0.9700
C15—H15B	0.9600	C22—H22B	0.9700
C15—H15C	0.9600	C32—H32A	0.9600
C5—C6	1.541 (6)	C32—H32B	0.9600
C5—H5A	0.9700	C32—H32C	0.9600
C5—H5B	0.9700	C31—H31A	0.9600
C6—H6A	0.9700	C31—H31B	0.9600
C6—H6B	0.9700	C31—H31C	0.9600
C24—C32	1.504 (7)	C23—H23A	0.9700
C24—C23	1.524 (7)	C23—H23B	0.9700
C9—C8—C1	112.7 (3)	C27—C28—C26	60.4 (3)
C9—C8—C7	107.3 (3)	C27—C28—C34	120.5 (4)
C1—C8—C7	113.9 (3)	C26—C28—C34	119.8 (4)
C9—C8—H8	107.5	C27—C28—C29	116.9 (3)
C1—C8—H8	107.5	C26—C28—C29	115.3 (3)
C7—C8—H8	107.5	C34—C28—C29	113.8 (4)
C15—C7—C6	109.9 (3)	C3—C14—H14A	109.5
C15—C7—C16	108.1 (3)	C3—C14—H14B	109.5
C6—C7—C16	104.3 (3)	H14A—C14—H14B	109.5
C15—C7—C8	107.9 (3)	C3—C14—H14C	109.5
C6—C7—C8	113.6 (3)	H14A—C14—H14C	109.5
C16—C7—C8	112.9 (3)	H14B—C14—H14C	109.5
C2—C1—C13	117.6 (3)	C30—C18—C19	115.9 (3)
C2—C1—C8	121.0 (3)	C30—C18—C25	112.6 (3)
C13—C1—C8	112.7 (3)	C19—C18—C25	122.4 (3)
C2—C1—C3	59.2 (2)	C30—C18—C20	118.1 (3)
C13—C1—C3	119.0 (3)	C19—C18—C20	58.7 (3)
C8—C1—C3	117.6 (3)	C25—C18—C20	119.4 (4)
C3—C2—C1	61.2 (2)	C19—C20—C31	119.0 (4)
C3—C2—Cl1	119.8 (3)	C19—C20—C21	117.5 (4)
C1—C2—Cl1	119.2 (3)	C31—C20—C21	113.2 (4)
C3—C2—Cl2	118.2 (3)	C19—C20—C18	60.4 (3)
C1—C2—Cl2	123.1 (3)	C31—C20—C18	120.8 (4)
Cl1—C2—Cl2	108.67 (19)	C21—C20—C18	116.3 (4)
C10—C9—C11	60.0 (2)	C27—C26—C28	58.9 (3)
C10—C9—C8	130.8 (3)	C27—C26—C25	130.2 (3)
C11—C9—C8	122.9 (3)	C28—C26—C25	124.3 (3)
C10—C9—H9	111.2	C27—C26—H26	111.2
C11—C9—H9	111.2	C28—C26—H26	111.2
C8—C9—H9	111.2	C25—C26—H26	111.2
C9—C10—C11	61.0 (2)	C28—C34—H34A	109.5
C9—C10—Br1	126.2 (3)	C28—C34—H34B	109.5
C11—C10—Br1	121.4 (3)	H34A—C34—H34B	109.5
C9—C10—Br2	114.6 (3)	C28—C34—H34C	109.5

C11—C10—Br2	117.5 (3)	H34A—C34—H34C	109.5
Br1—C10—Br2	109.24 (18)	H34B—C34—H34C	109.5
C1—C13—C12	112.8 (3)	C29—C30—C18	112.5 (3)
C1—C13—H13A	109.0	C29—C30—H30A	109.1
C12—C13—H13A	109.0	C18—C30—H30A	109.1
C1—C13—H13B	109.0	C29—C30—H30B	109.1
C12—C13—H13B	109.0	C18—C30—H30B	109.1
H13A—C13—H13B	107.8	H30A—C30—H30B	107.8
C11—C12—C13	113.5 (3)	C11—C17—H17A	109.5
C11—C12—H12A	108.9	C11—C17—H17B	109.5
C13—C12—H12A	108.9	H17A—C17—H17B	109.5
C11—C12—H12B	108.9	C11—C17—H17C	109.5
C13—C12—H12B	108.9	H17A—C17—H17C	109.5
H12A—C12—H12B	107.7	H17B—C17—H17C	109.5
C2—C3—C4	117.7 (3)	C22—C21—C20	113.2 (5)
C2—C3—C14	119.4 (3)	C22—C21—H21A	108.9
C4—C3—C14	113.1 (3)	C20—C21—H21A	108.9
C2—C3—C1	59.6 (2)	C22—C21—H21B	108.9
C4—C3—C1	117.0 (3)	C20—C21—H21B	108.9
C14—C3—C1	120.1 (3)	H21A—C21—H21B	107.7
C10—C11—C17	119.4 (4)	C24—C33—H33A	109.5
C10—C11—C12	116.7 (3)	C24—C33—H33B	109.5
C17—C11—C12	114.9 (4)	H33A—C33—H33B	109.5
C10—C11—C9	59.0 (2)	C24—C33—H33C	109.5
C17—C11—C9	119.6 (4)	H33A—C33—H33C	109.5
C12—C11—C9	116.1 (3)	H33B—C33—H33C	109.5
C3—C4—C5	111.5 (4)	C20—C19—C18	60.9 (3)
C3—C4—H4A	109.3	C20—C19—Cl4	120.3 (3)
C5—C4—H4A	109.3	C18—C19—Cl4	119.5 (3)
C3—C4—H4B	109.3	C20—C19—Cl3	118.9 (3)
C5—C4—H4B	109.3	C18—C19—Cl3	122.5 (3)
H4A—C4—H4B	108.0	Cl4—C19—Cl3	108.4 (2)
C7—C16—H16A	109.5	C28—C27—C26	60.7 (3)
C7—C16—H16B	109.5	C28—C27—Br3	123.0 (3)
H16A—C16—H16B	109.5	C26—C27—Br3	124.0 (3)
C7—C16—H16C	109.5	C28—C27—Br4	117.8 (3)
H16A—C16—H16C	109.5	C26—C27—Br4	115.5 (3)
H16B—C16—H16C	109.5	Br3—C27—Br4	108.9 (2)
C7—C15—H15A	109.5	C30—C29—C28	113.1 (4)
C7—C15—H15B	109.5	C30—C29—H29A	109.0
H15A—C15—H15B	109.5	C28—C29—H29A	109.0
C7—C15—H15C	109.5	C30—C29—H29B	109.0
H15A—C15—H15C	109.5	C28—C29—H29B	109.0
H15B—C15—H15C	109.5	H29A—C29—H29B	107.8
C4—C5—C6	112.8 (3)	C21—C22—C23	113.8 (4)
C4—C5—H5A	109.0	C21—C22—H22A	108.8
C6—C5—H5A	109.0	C23—C22—H22A	108.8
C4—C5—H5B	109.0	C21—C22—H22B	108.8
C6—C5—H5B	109.0	C23—C22—H22B	108.8

H5A—C5—H5B	107.8	H22A—C22—H22B	107.7
C5—C6—C7	120.3 (4)	C24—C32—H32A	109.5
C5—C6—H6A	107.2	C24—C32—H32B	109.5
C7—C6—H6A	107.2	H32A—C32—H32B	109.5
C5—C6—H6B	107.2	C24—C32—H32C	109.5
C7—C6—H6B	107.2	H32A—C32—H32C	109.5
H6A—C6—H6B	106.9	H32B—C32—H32C	109.5
C32—C24—C23	113.1 (5)	C20—C31—H31A	109.5
C32—C24—C33	105.4 (4)	C20—C31—H31B	109.5
C23—C24—C33	102.6 (4)	H31A—C31—H31B	109.5
C32—C24—C25	108.7 (4)	C20—C31—H31C	109.5
C23—C24—C25	114.3 (3)	H31A—C31—H31C	109.5
C33—C24—C25	112.4 (4)	H31B—C31—H31C	109.5
C26—C25—C18	112.8 (3)	C24—C23—C22	119.2 (5)
C26—C25—C24	106.7 (3)	C24—C23—H23A	107.5
C18—C25—C24	113.8 (3)	C22—C23—H23A	107.5
C26—C25—H25	107.8	C24—C23—H23B	107.5
C18—C25—H25	107.8	C22—C23—H23B	107.5
C24—C25—H25	107.8	H23A—C23—H23B	107.0